

TWO NOVEL HUMULENOIDS FROM TORILIS SCABRA D.C.

Hideji ITOKAWA,* Hajime MATSUMOTO, Susumu MIHASHI, and Yoichi IITAKA†
Tokyo College of Pharmacy,

1432-1 Horinouchi, Hachioji, Tokyo 192-03

† Faculty of Pharmaceutical Sciences, The University of Tokyo,
Hongo, Bunkyo-ku, Tokyo 113

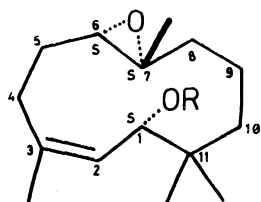
Two humulene type sesquiterpenoids were isolated from Torilis scabra D.C. and their structures were determined by X-ray analysis.

Torilis scabra D.C. (Japanese name "O-yabujirami"; *Umberifellae*) is a biennial and widely distributing only in Japan and south Korea. From the fruits, caucalol diacetate and apocaucalol diacetate were isolated by Mitsui¹⁾ and the structure of the former was determined by Sasaki and co-workers²⁾ as indicated in Chart 1. In the course of our study on a chinese crude drug "She-Hunzi", we have reported the isolation of germacrene, eudesmane, oppositane, and the novel cycloeudesmane type sesquiterpenoids from Torilis japonica D.C. (Japanese name "Yabujirami").³⁾ In this paper, we present the isolation of two novel humulenooids from T. scabra.

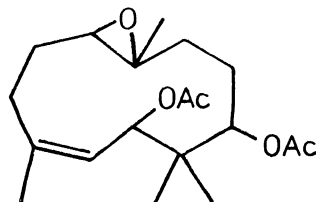
The fresh fruits of T. scabra were extracted in the same manner,³⁾ and the hexane extract was chromatographed on silica gel repeatedly to afford compound 1 and 2.⁴⁾ Compound 1 (C₁₇H₂₈O₃) was an acetyl ester of sesquiterpene from NMR spectra [¹H-; 2.03 3H, s; ¹³C-; 170.4 (s)]. Compound 2 (C₁₅H₂₆O₂) was assigned as a monoacetyl derivative of 1 because base catalyzed hydrolysis of 1 gave 2.

Compound 2 was converted to a p-bromobenzoyl ester (3), and the CD spectra of 3 showed the negative Cotton curve (max 246 nm; Δε = -13.6 in EtOH). The absolute configuration of 3 recrystallized from pet. ether (Mp. 175-179°C) was determined by X-ray method using bromine as an anomalous scatterer as illustrated in Fig. 1.

Crystal data: C₂₂H₂₉O₃Br, orthorhombic, P2₁2₁2₁; Z=4, a=8.435, b=40.229, c=6.239 (Å). A total of 1451 reflections were recorded using a Philips four-cir-



1: R=Ac
2: R=H
3: R=p-Br-C₆H₄-CO



Caucalol diacetate

Chart 1.

cle diffractometer with graphite-monochromated Cu-K α radiation. The structure was solved by heavy atom method. Block-matrix least-squares refinement with anisotropic temperature factors for non-hydrogen atoms, and isotropic temperature factors for hydrogen atoms converged to a conventional R-value of 0.06.

Although the conformation of 3 in the solution was not clear, S configuration at C1 position, the coupling constant of the C1 proton ($J=10$ Hz), and the negative Cotton curve exhibit that the conformation around the allylic position of 3 should be shown as 4. The chiroptical method for determining the absolute configuration of allylic alcohol by Harada⁵⁾ is applicable to that of eleven-membered ring.

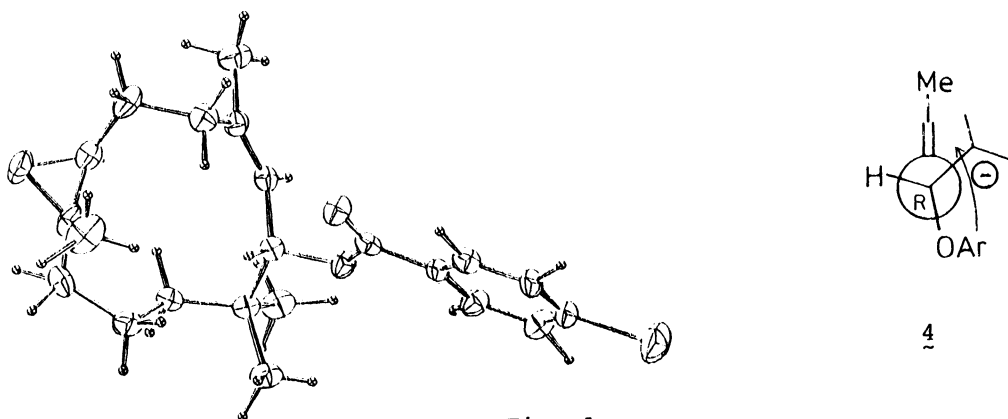


Fig. 1.

References

- 1) S. Mitsui, *Rika Gaku Kenkyusho Hokoku*, 20, 529, 533, 540, 549, 695, 708, 715 (1941).
- 2) S. Sasaki, Y. Itagaki, H. Moriyama, K. Nakanishi, E. Watanabe, and T. Aoyama, *Tetrahedron Lett.*, 1966, 623.
- 3) H. Itokawa, H. Matsumoto, and S. Mihashi, *Chem. Pharm. Bull.*, 31, 1743 (1983); *Chem. Lett.*, 1983, 1253.
- 4) Compound 1 [α]_D +21.4^o (c 0.36, CHCl₃). Mp 125-130^oC. IR (CCl₄): 3000, 2970, 2930, 2870, 1730, 1655. ¹H-NMR (CDCl₃): 0.86(3H,s), 0.94(3H,s), 1.36(3H,s), 1.82(3H,s), 2.03(3H,s), 2.84(1H,dd,J=4,11 Hz), 5.28(1H,d,J=10 Hz), 5.54(1H,d,J=10 Hz). ¹³C-NMR (CDCl₃): 16.66(q), 19.72(t), 21.16(q), 22.72(q), 23.01(q), 24.04(q), 25.02(t), 29.12(t), 37.02(s), 37.59(t), 38.69(t), 60.60(s), 60.60(d), 73.40(d), 122.58(d), 140.86(s), 170.44(s).
Compound 2 [α]_D +30.9^o (c 0.22, CHCl₃). Mp 125-130^oC. IR (CCl₄): 3630, 2940, 2920, 2870, 1660, ¹H-NMR (CDCl₃): 0.88(3H,s), 0.98(3H,s), 1.32(3H,s), 1.84(3H,s), 2.87(1H,dd,J=4,11 Hz), 4.30(1H,d,J=11 Hz), 5.41(1H,br-d,J=11 Hz). ¹³C-NMR (CDCl₃): 16.55(q), 19.66(t), 22.89(q), 22.89(q), 25.02(q), 25.02(t), 28.71(t), 37.42(s), 37.54(t), 39.03(t), 60.54(s), 60.54(d), 70.86(d), 126.90(d), 138.03(s).
- 5) N. Harada, J. Iwabuchi, Y. Yokota, H. Uda, and K. Nakanishi, *J. Am. Chem. Soc.*, 103, 5590, (1981); N. C. Gonnella, K. Nakanishi, V. S. Martin, and K. B. Sharpless, *ibid.*, 104, 3775 (1982).

(Received July 13, 1983)